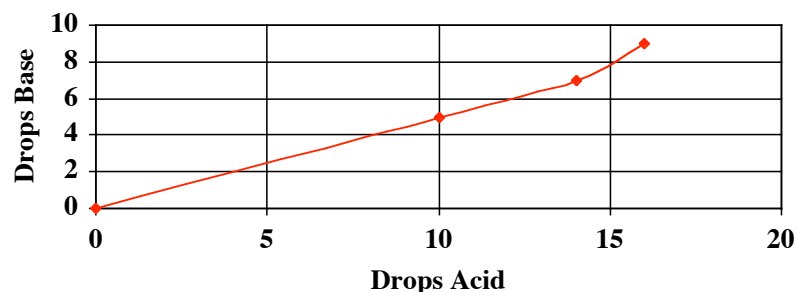


## ACID BASE NEUTRALIZATION



### CONCLUSIONS Paragraph

Topic Sentence (answers question.....)

Explain/ Describe results in all trials

Describe /Explain graph: slope +/-, prediction for 20 drops of acid

Answer to question:” Which was stronger the acid or the base? Use textbook page 190 to answer.

Conclusion statement: sum up what you said

Compound	Charge	Symbol	pH Range	Color in Indicator
<b>ACID</b>				
<b>BASE</b>				
<b>WATER</b>				

If you mix an acid and a base you can make an neutral solution.

### NEUTRALIZATION LAB

QUESTION: How much base is needed to neutralize an acid?

HYPOTHESIS: I predict that....

DATA:

TRIAL #	Drops ACID	Drops BASE
1		
2		
3		

METHOD:

1. 5 mL H<sub>2</sub>O. 5 drops indicator. COLOR: \_\_\_\_\_
2. 10 drops ACID. COLOR: \_\_\_\_\_
3. Drops base until original color.  
Drops acid: \_\_\_\_\_ Drops Base : \_\_\_\_\_
4. Repeat.. 5 drops acid